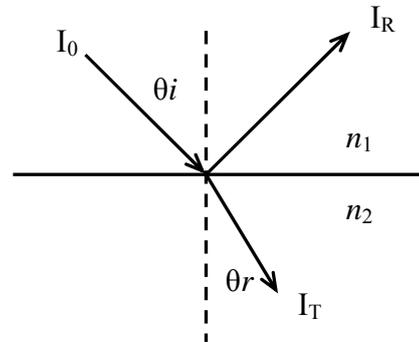


Brewster Angle Microscopy

1. Principle of The Technique

I_0 : intensity of incident light
 I_R : reflected intensity of light
 I_T : transmitted intensity of light



$$n_1 \sin \theta_i = n_2 \sin \theta_r$$

- I_R is function of : **incidence angle, θ_i ;**
polarization of the light;
details of interface.
- Reflectivity: $R = \frac{I_R}{I_o}$
- for p-polarized incident beam (electric field in the plane of incidence)

$$R_p = \left[\frac{\tan(\theta_i - \theta_r)}{\tan(\theta_i + \theta_r)} \right]^2 \dots\dots\dots(1)$$

for S-polarized incident beam (electric field perpendicular to the plane of incidence):

$$R_s = \left[\frac{\sin(\theta_i - \theta_r)}{\sin(\theta_i + \theta_r)} \right]^2 \dots\dots\dots(2)$$

- R_s and R_p are functions of incident θ_i .

When the light is incident from air ($n_1=1$) to water ($n_2=1.33$)

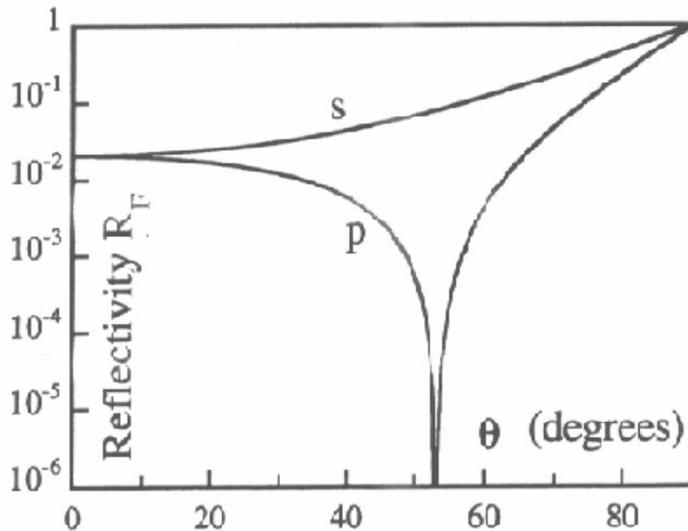


Fig. 1 The reflectivities R_s^F and R_p^F as a function of the incidence angle θ_i for a Fresnel interface between two media of refractive indices $n_1 = 1$ and $n_2 = 1.33$.

- At an angle of incident θ_B (called Brewster Angle), $R_p \rightarrow 0$

from Eq.(1), $\tan(\theta_i + \theta_r) = \infty$

$\rightarrow (\theta_i + \theta_r) = \pi/2$, with assistance of ($n_1 \sin \theta_i = n_2 \sin \theta_r$)

$$\rightarrow \tan \theta_B = \frac{n_2}{n_1}$$

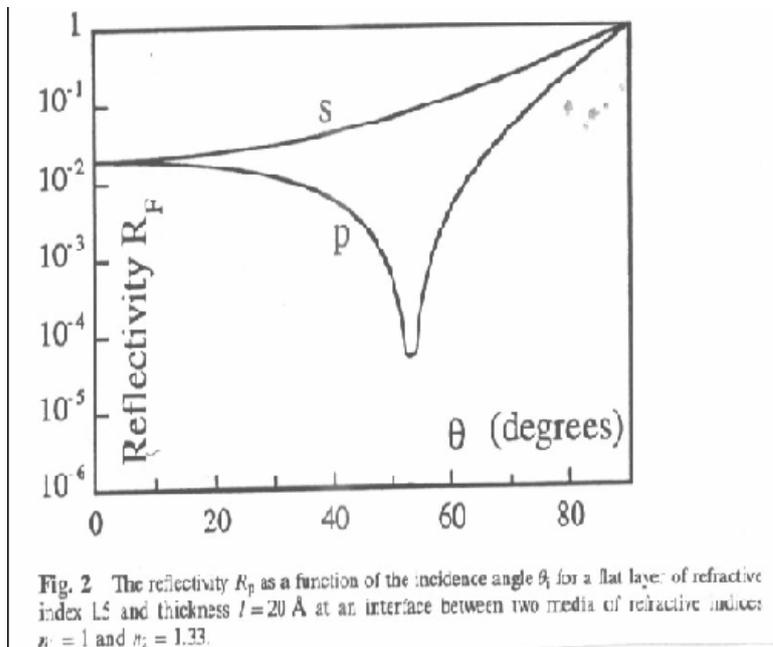
for air /water, $\theta_B = 53.06^\circ$

- For real interface, R_p decrease to a minimum value at Brewster angle but does not vanish. (Due to the discrepancy between an ideal interface and a real interface)

➤ Real interface

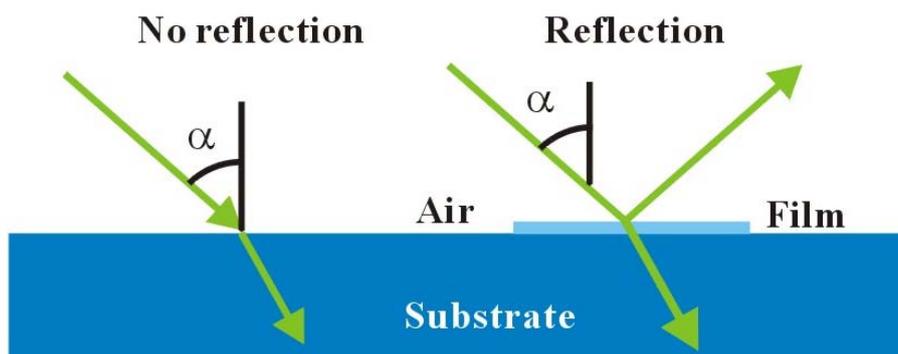
- \rightarrow the refractive index varies smoothly from n_1 to n_2 , thickness ℓ .
- \rightarrow rough (not exact uniform)
- \rightarrow optical anisotropic (due to preferential orientation of molecules in the interface)

- for a surfactant monolayer at water surface, $n \sim 1.4$, $l \sim 20\text{\AA}$,
 $\rightarrow R_p(\theta_B) \approx 5 \times 10^{-5}$ (see figure below)

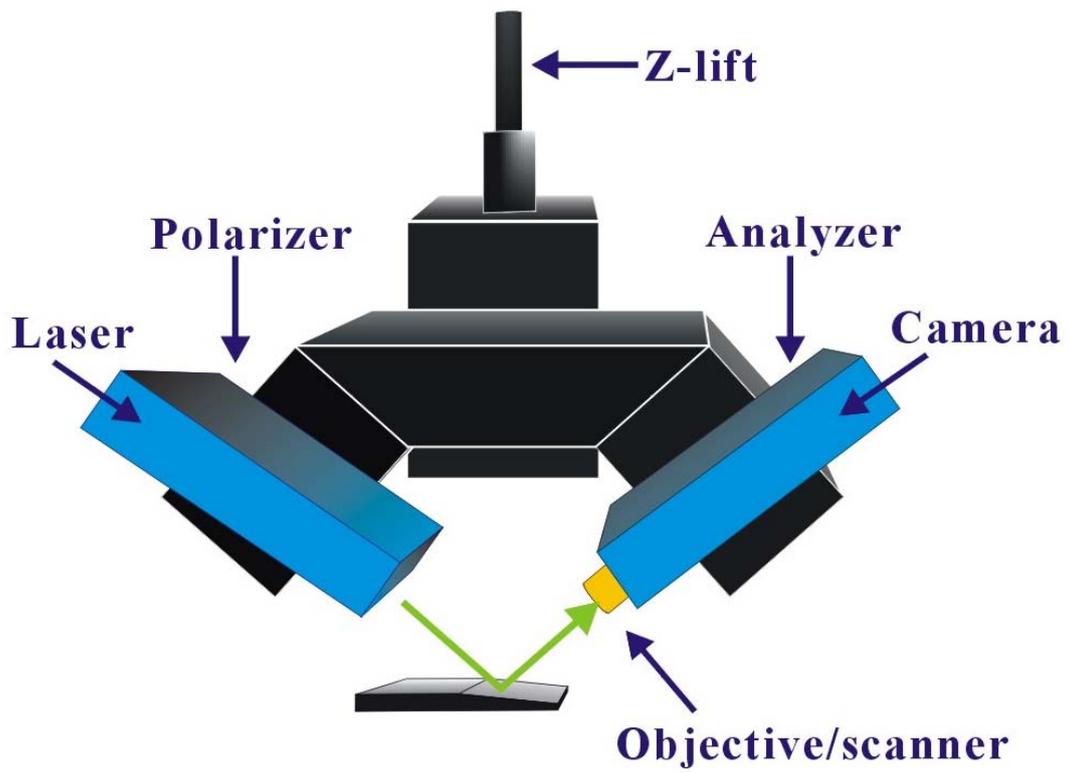


$\rightarrow R_p$ is very low in a narrow range of θ_i near the Brewster angle (θ_B), which is different from that of an ideal interface only in the vicinity of θ_B .

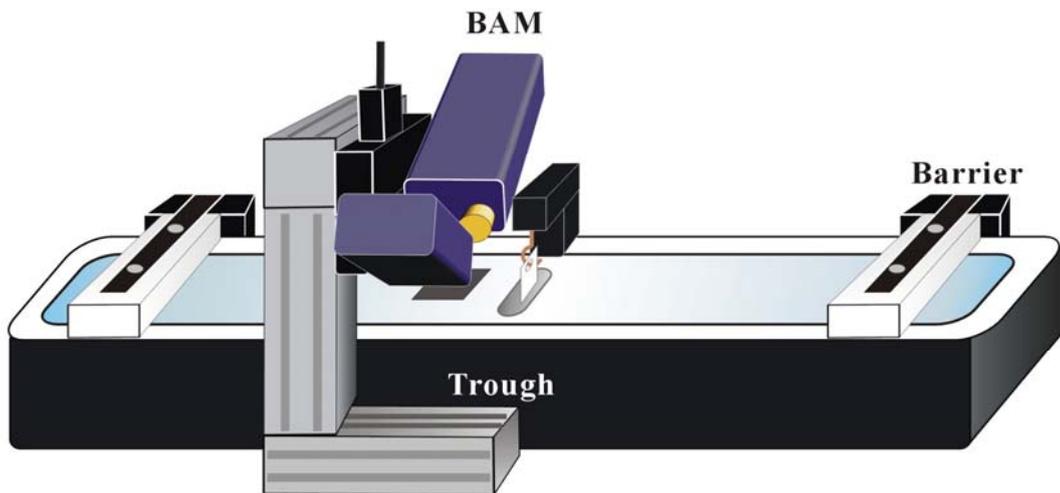
■ Equipments of BAM

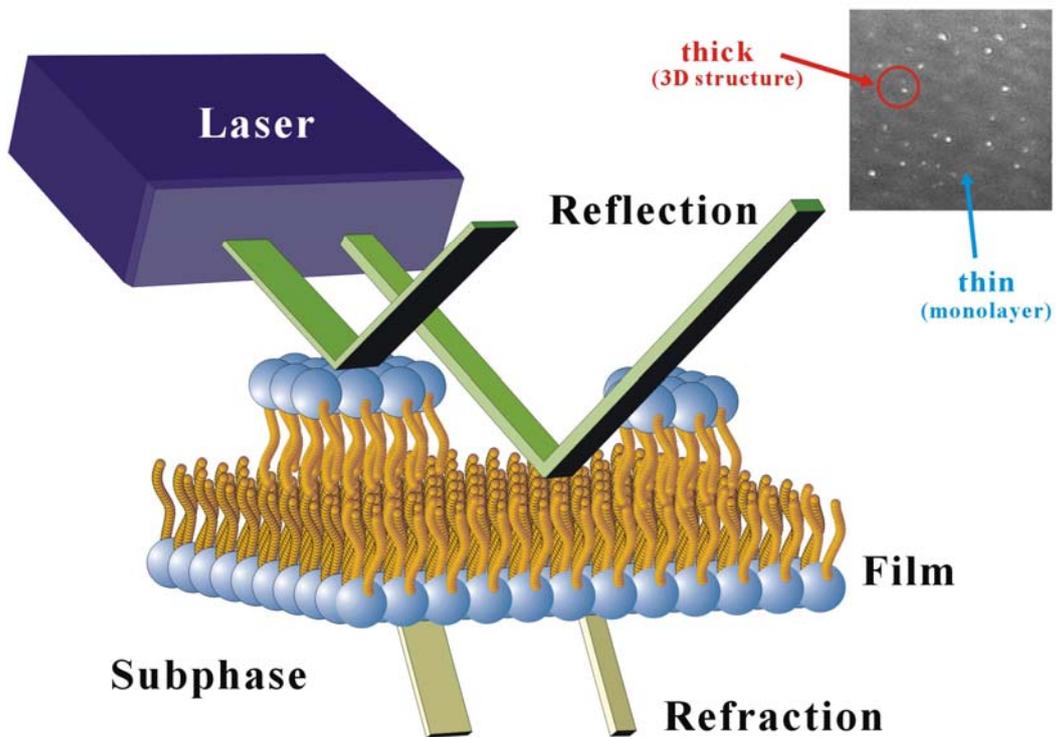
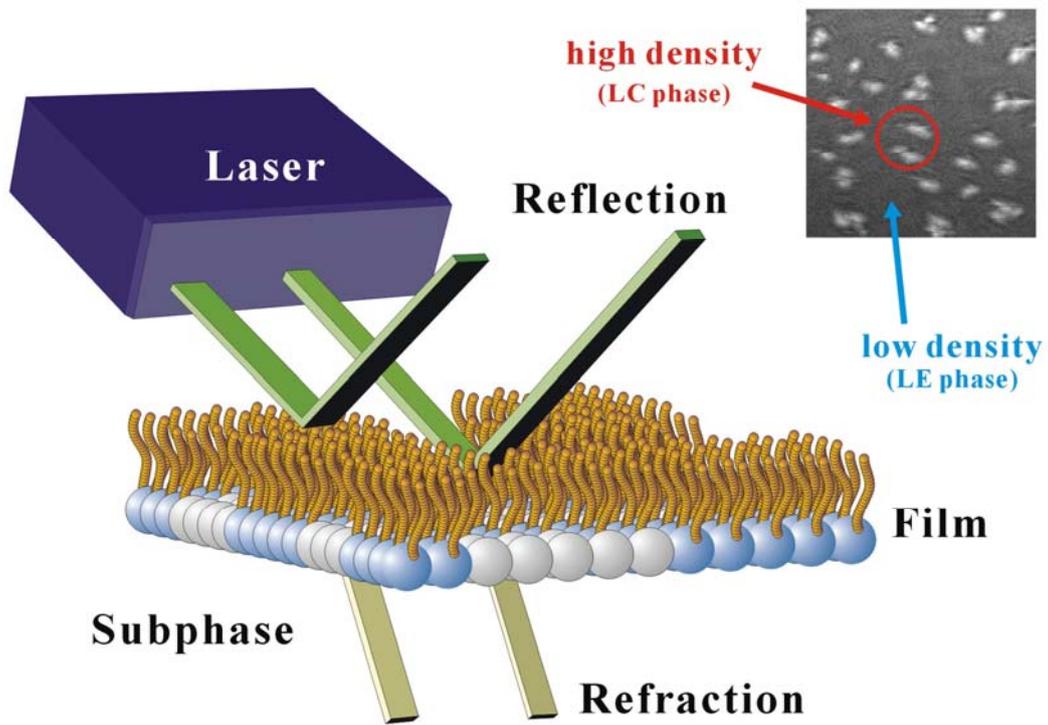


$$\tan \alpha = \frac{n_{\text{substrate}}}{n_{\text{air}}}$$

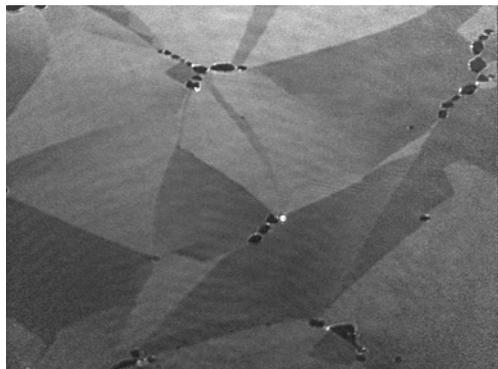
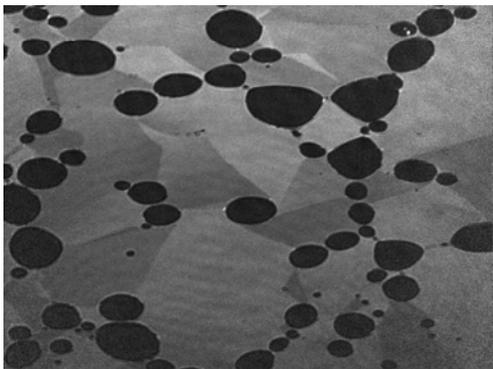
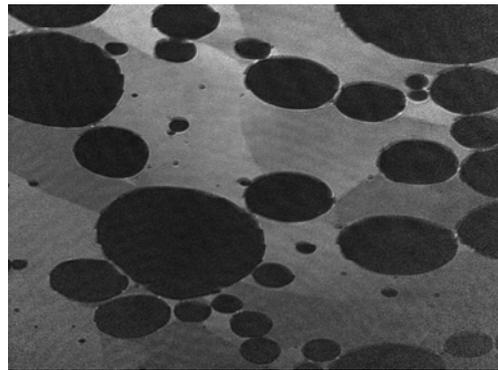
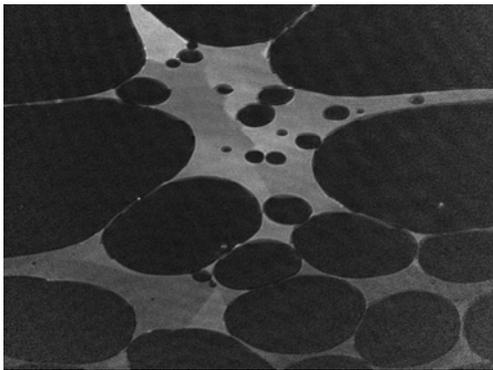


■ BAM on trough



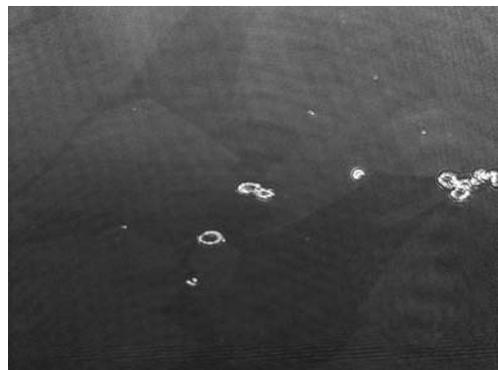


■ BAM images of stearic acid at gas phases ($\pi \approx 0$)



$\pi \approx 5$ ($A=24.1$)

027, $\pi \approx 14.8$ ($A \approx 22.1$)

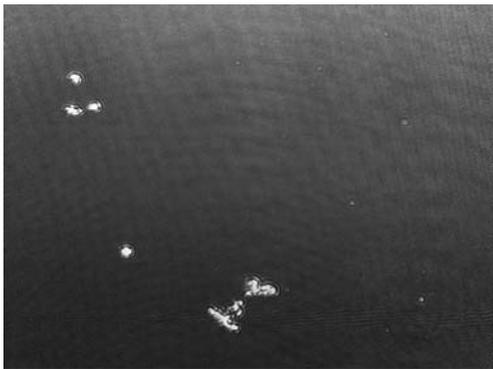


$\pi \approx 21.8$ ($A \approx 20.6$)

$\pi \approx 30$ ($A \approx 20.5$)



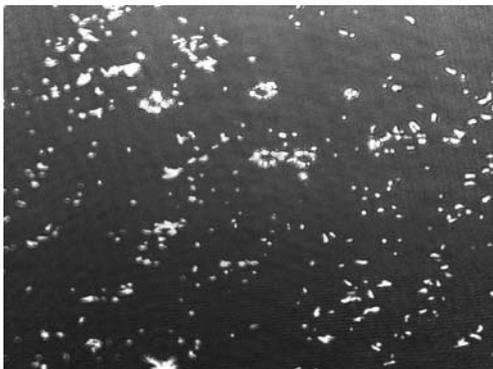
2035, $A=19.5$, $\pi=40$



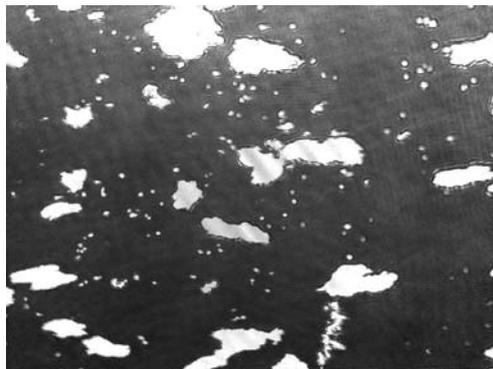
2036, $\pi=44$



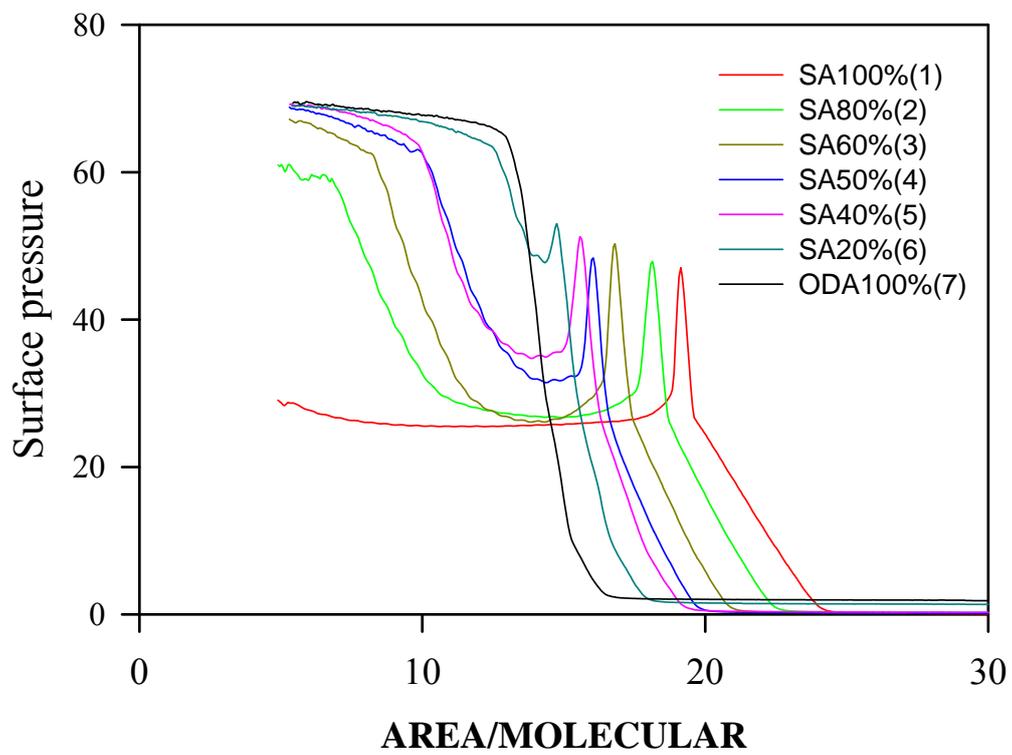
2040, $\pi=25$



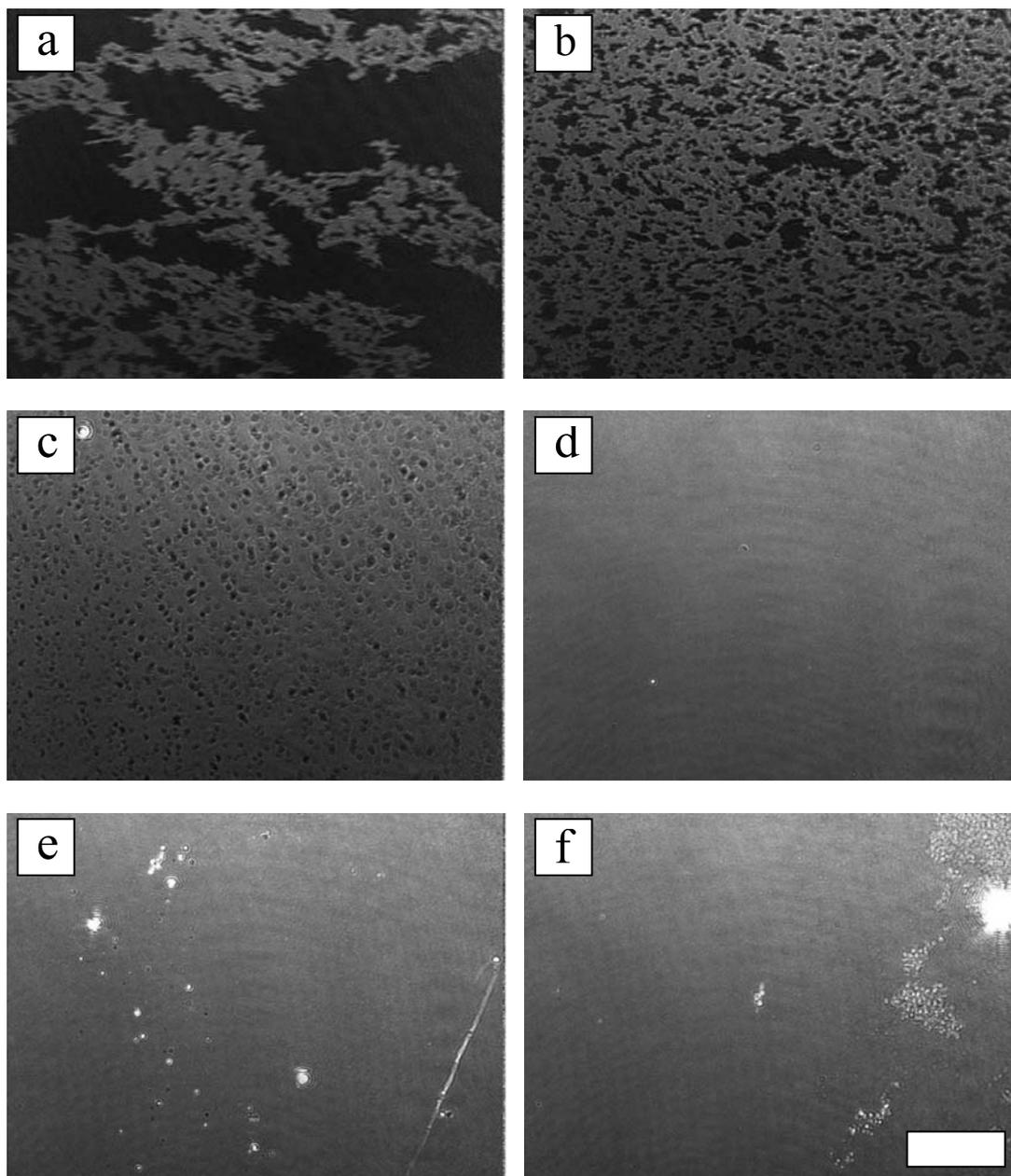
2054, $\pi=19$



SA先打等温線

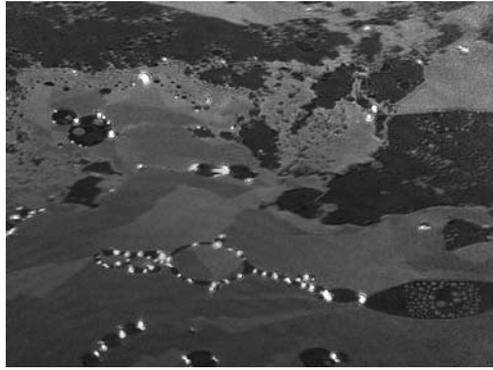
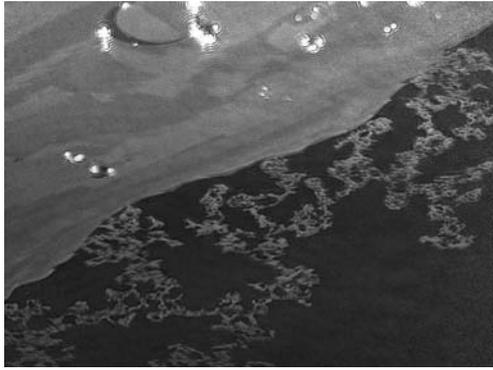


BAM of ODA (octadecylamine, C₁₈-NH₂)



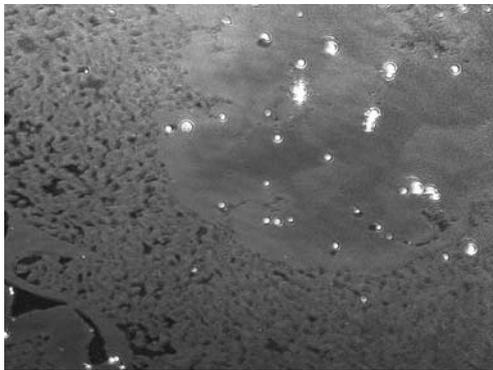
BAM images of pure ODA monolayer on air/ water interface at 25 °C. The images correspond to states of (a): $A=24.7 \text{ \AA}^2/\text{molecule}$, $\pi=0 \text{ mN/m}$; (b): $A=22.5 \text{ \AA}^2/\text{molecule}$, $\pi=0.3 \text{ mN/m}$; (c): $A=20.2 \text{ \AA}^2/\text{molecule}$, $\pi=1.6 \text{ mN/m}$; (d): $\pi=61.5 \text{ mN/m}$; (e): $\pi=63.8 \text{ mN/m}$ near collapse point; (f): $\pi=65.0 \text{ mN/m}$ after collapse point.

SA 40% BAM images: (1). 012, A=32.9, PI=0.2; (2)040, A=28.1, PI=0.3;



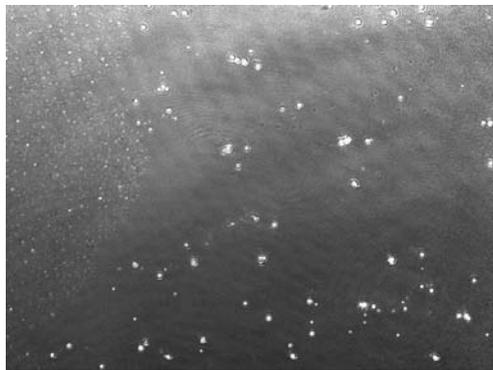
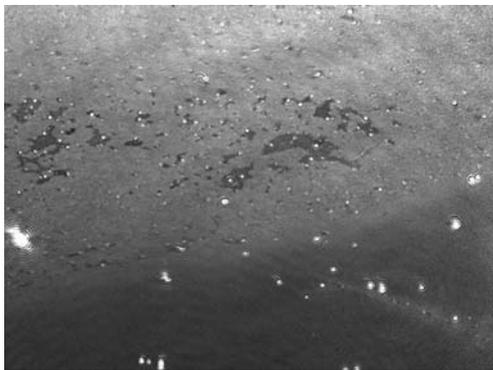
(3). 054, A=26.4, PI=3.9

(4) 065, A=24.5, PI=14



(5). 070, A= , PI=18.8

(6).074, A= , PI=30.4

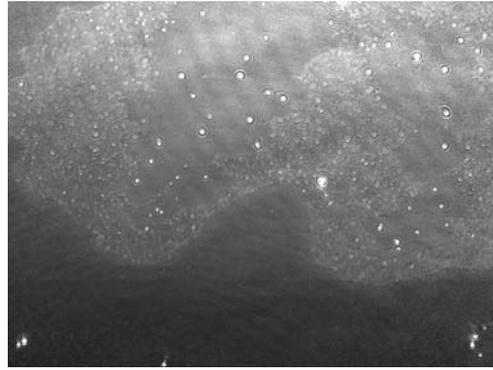
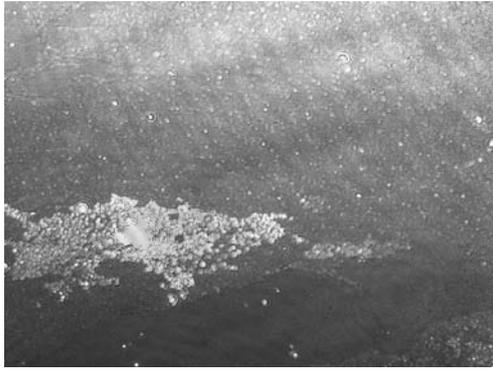


(7). 078, A= , PI=41.8

(8) 081, A= , PI= 47.2 collapse point

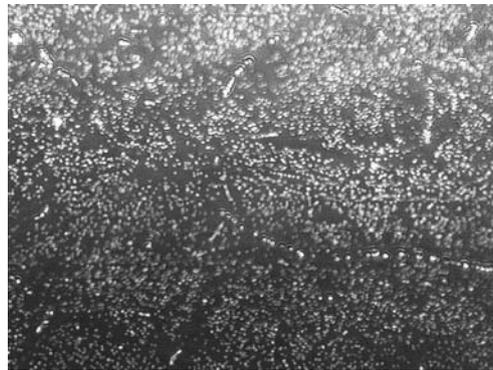
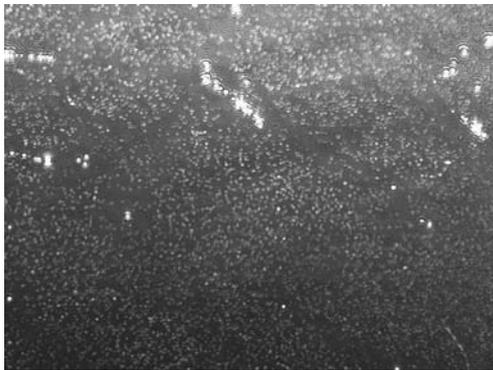


(9) 085, PI= 37.7, after collapse (10) 093, PI=33 (21.3), after collapse, 2 phases



(11) 097, PI=33.9 (A=20.7)

(12) 101, PI=35.5 (A=20.3)



038, A=28.3, PI=0.3

